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## Key indicators

Single-crystal X-ray study  
 $T = 100$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.038  
 $wR$  factor = 0.102  
Data-to-parameter ratio = 25.4For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.High-spin  $[\text{Fe}(\text{TTP})(\text{THF})_2]$ 

The title compound, bis(tetrahydrofuran)(5,10,15,20-tetra-*p*-tolylporphyrinato)iron(II),  $[\text{Fe}(\text{C}_{48}\text{H}_{36}\text{N}_4)(\text{C}_4\text{H}_8\text{O})_2]$  or  $[\text{Fe}(\text{TTP})(\text{THF})_2]$  (TTP = dianion of *meso*-tetratolylporphyrin), is a six-coordinate high-spin iron(II) tetratolylporphyrin with two tetrahydrofuran molecules as axial ligands. The  $\text{Fe}^{\text{II}}$  atom is at an inversion center. The average  $\text{Fe}-\text{N}$  distance is 2.067 (7) Å and the  $\text{Fe}-\text{O}$  distance is 2.3208 (8) Å.

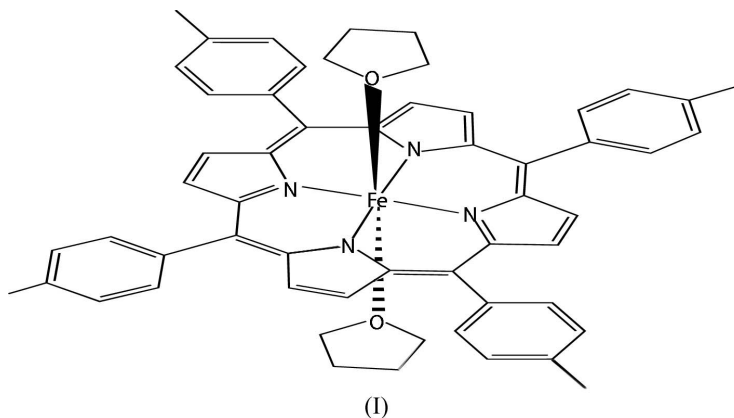
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## Comment

Reed *et al.* (1980) reported the structure of the first six-coordinate high-spin iron(II) porphyrinate,  $[\text{Fe}(\text{TPP})(\text{THF})_2]$ . Subsequently, Lecomte *et al.* (1986) performed an electron-density study of the compound. In this paper, we report the molecular structure of another six-coordinate high-spin iron(II) porphyrinate, the title compound, (I),  $[\text{Fe}(\text{TTP})(\text{THF})_2]$ .



The molecular structure of (I), with the atom-numbering scheme, is shown in Fig. 1. The  $\text{Fe}^{\text{II}}$  atom is located at a crystallographic inversion center. As a result, the  $\text{Fe}^{\text{II}}$  atom is centered in the porphyrin plane. The independent  $\text{Fe}-\text{N}$  bond distances are 2.0619 (9) and 2.0713 (9) Å, which are similar to those in  $[\text{Fe}(\text{TPP})(\text{THF})_2]$  (2.054 and 2.060 Å; Reed *et al.*, 1980). These values are substantially longer than the value of 1.996 (6) Å in the low-spin six-coordinate iron(II) porphyrinate  $[\text{Fe}(\text{TPP})(\text{THT})_2]$  (THT is tetrahydrothiophene; Mashiko *et al.*, 1979).

The axial  $\text{Fe}-\text{O}$  bond distance [2.3208 (8) Å] is similar to the value of 2.351 (3) Å in  $[\text{Fe}(\text{TPP})(\text{THF})_2]$  (Reed *et al.*, 1980). This bond distance indicates that the THF ligand is a weak-field ligand. The  $\text{Fe}-\text{O}$  vector is tipped by 1.3 (1)° from the normal to the mean plane through the plane of the four N atoms. This is smaller than the 4.3° tip seen in

[Fe(TPP)(THF)<sub>2</sub>] (Reed *et al.*, 1980). This deviation also breaks the  $D_{4h}$  symmetry of the FeN<sub>4</sub>O<sub>2</sub> coordination. The THF ligand has an envelope configuration with approximate  $C_s$  symmetry. The O atom is displaced from the mean plane of the four C atoms by 0.56 (3) Å.

## Experimental

All experimental manipulations were performed under a purified nitrogen atmosphere using standard Schlenk techniques. [Fe(TTP)]<sub>2</sub>O (35 mg) was reduced in tetrahydrofuran (10 ml) in the presence of excess EtSH overnight. The solution was transferred into glass tubes and layered with hexane. A week later, high-quality red crystals of (I) formed.

### Crystal data

[Fe(C <sub>48</sub> H <sub>36</sub> N <sub>4</sub> )(C <sub>4</sub> H <sub>8</sub> O) <sub>2</sub> ]	$D_x = 1.312 \text{ Mg m}^{-3}$
$M_r = 868.87$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 7064 reflections
$a = 13.6820$ (2) Å	$\theta = 2.5\text{--}31.5^\circ$
$b = 9.7320$ (1) Å	$\mu = 0.39 \text{ mm}^{-1}$
$c = 17.1014$ (2) Å	$T = 100$ (2) K
$\beta = 104.964$ (1) $^\circ$	Needle, red
$V = 2199.89$ (5) Å <sup>3</sup>	$0.49 \times 0.15 \times 0.09 \text{ mm}$
$Z = 2$	

### Data collection

Bruker SMART APEX2 CCD area-detector diffractometer	7311 independent reflections
$\varphi$ and $\omega$ scans	6283 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 2004)	$R_{\text{int}} = 0.028$
$T_{\text{min}} = 0.83$ , $T_{\text{max}} = 0.97$	$\theta_{\text{max}} = 31.5^\circ$
31 446 measured reflections	$h = -20 \rightarrow 17$
	$k = -11 \rightarrow 14$
	$l = -22 \rightarrow 25$

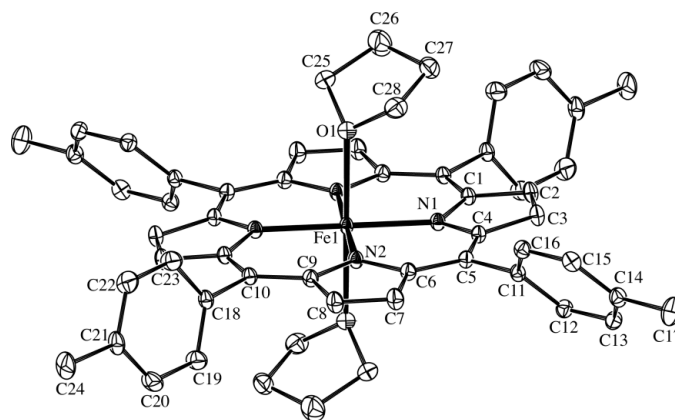
### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0487P)^2 + 1.0408P]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.102$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.53 \text{ e \AA}^{-3}$
7311 reflections	$\Delta\rho_{\text{min}} = -0.31 \text{ e \AA}^{-3}$
288 parameters	
H-atom parameters constrained	

**Table 1**

Selected geometric parameters (Å,  $^\circ$ ).

Fe1—N1	2.0619 (9)	Fe1—O1	2.3208 (8)
Fe1—N2	2.0713 (9)		
N1—Fe1—N2	89.57 (4)	N2—Fe1—O1	88.96 (3)
N1—Fe1—O1	89.34 (3)		



**Figure 1**

The molecular structure of (I), showing 50% probability displacement ellipsoids. H atoms have been omitted. Unlabeled atoms are related by the symmetry operation  $-x, 1 - y, -z$ .

H atoms were positioned geometrically and treated as riding, with C—H distances in the range 0.95–0.99 Å and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: APEX2 (Bruker Nonius, 2004); cell refinement: APEX2 and SAINT (Bruker Nonius, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

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